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NEW RUBBER LABORATORY IN THE AMAZON VALLEY

Norman Bekkedahl, of the Bureau's rubber section, has just returned to Washington after spending 2 years in Brazil. In 1942, at the request of the Brazilian Government, the Bureau loaned the services of Dr. Bekkedahl to direct the establishment of a modern rubber laboratory at the Instituto Agronomico do Norte located in Belem in the State of Para (Technical News Bulletins 308, 316, and 321). He reports that the laboratory is now completely equipped and in operation, while its staff continues to be built up. Dr. Bekkedahl was assisted in this work by Fredrick L. Downs, one-time member of the Bureau staff and later an employee of the American Steel & Wire Co., Worcester, Mass., which granted him a leave of absence.

The Instituto Agronomico do Norte is one of several research institutes supported by the Brazilian Government and is somewhat similar to the Regional Research Laboratories of the United States Department of Agriculture. Dr. Fellsberto Cardoso de Camargo, the director of the Instituto, has concentrated the research program largely on rubber, since it is the reseat invertent events of

the Amazon region. The location of the Instituto at Belem is particularly fortunate, since all the rubber from the Amazon valley passes through this port at the mouth of the river. Whatever the relative positions of natural and synthetic rubber in the post-war years, the Amazon Valley must be considered as one of the more important potential sources of natural rubber.

The fact that Belem is a seaport and an important stop on South American and trans-Atlantic airlines also has made it possible for the Instituto to operate a so-called "factory" located in the jungle about 5 miles from the Instituto. Here about 100 liters of rubber latex are collected each day from wild trees and processed into raw rubber sheets.

A wide variety of types of trees, other than the common rubber tree (Hevea Brasiliensis) are under investigation. Studies of different methods of coagulation can be made under various typically primitive conditions. The rubber that is not used in experiments is sold through normal commercial channels.

Department of Agriculture. Dr. Fellsberto Cardoso de Camargo, the director of the Instituto, has concentrated the research program largely on rubber, since it is the most important export of States for the development of improved

¹ Published with approval of the Director of the Budget.

methods of testing and grading natural rubber, and for fundamental research on rubber of different origin and treatment.

ANALYSIS OF BUTADIENE AND STYRENE BY MASS SPECTROM-ETER

The Consolidated Engineering Corporation's mass spectrometer, recently installed at the Bureau, is being employed by A. Keith Brewer in analyses of liquid and gaseous materials for making synthetic rubber. This is an elaborate and highly specialized instrument that makes possible a quick, complete, and accurate analysis of samples of butadiene, styrene, and other monomers submitted by plant operators through the Rubber Reserve Co. Since many of the constituents are closely related hydrocarbons, the analyses by conventional methods would be impossible, or at least would involve considerable research on the method to be applied to each constituent. Studies are likewise being made of the purity of butadiene and styrene produced in different plants and by different Other projects deal with the methods composition of recycled butadiene and styrene, which are recovered after polymerization for use in a later run. Such studies furnish information regarding the amount and kind of purification necessary for these materials. Analyses of vent gases have led to the conclusion that considerable amounts of valuable materials were being allowed to escape, and steps have been taken to reduce or eliminate the waste.

PREPARING REFRACTORY OXIDES, SILICATES, AND CERAMIC MATERIALS FOR ANALYSIS

As an outgrowth of work in preparing refractory platinum-metal alloys for analysis by treatment with acid mixtures at high temperatures, it has been found that numerous other substances that have caused trouble for the analyst will respond to a similar method of treatment. Aluminum oxide, beryllium oxide, tin oxide, cerium oxide, and thorium oxide, all of which are very resistant to the usual methods of chemical freatment if they have been ignited at high temperatures, dissolve rather readily in hydrochloric acid at temperatures ranging from 200° to 300° C. The treatment is applied in sealed glass tubes enclosed in steel pressure vessels charged with compressed carbon dioxide.

in sufficient amounts to keep the glass tubes from bursting under the high pressures developed by the acid (TNB331, November 1944).

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Most aluminum-bearing minerals, as well as a variety of other minerals and ceramic materials, can be decomposed, preparatory to analysis, by this treatment. The work is reported by Edward Wichers, William G. Schlecht, and Charles L. Gordon in the December number of the Journal of Research (RP1621).

GAGE FOR MEASURING VAPOR PRESSURES OF CORROSIVE LIQUIDS

The experimental work reported in TNB331 (November 1944) on the use of hydrochloric acid and other acid mixtures at temperatures up to 300° C to prepare for analysis refractory platinum-metal alloys and numerous refractory oxides, ceramic materials, and minerals, has necessarily involved the development of suitable laboratory technics for conducting the new method of treatment. It was also necessary to measure the high pressures developed by the acid mixtures under the conditions of operation. In RP1622 in the December number of the Journal of Research, Charles L. Gordon, William G. Schlecht, and Edward Wichers describe an allglass gage, with auxiliary equipment, suitable for the approximate measurement of the vapor-pressures of corrosive liquids. They report the pressures developed by various strengths of hydrochloric acid and its mixtures with nitric acid or perchloric acid at temperatures up to 300° C. The observed pressures ranged up to 4,000 lb/in 1

The paper describes how to construct, fill, and seal the glass tubes used as reaction vessels, and how to construct and use a simple steel pressure vessel to prevent the glass tubes from bursting. Carbon dioxide, added in solid form (dry ice) furnishes the compensating pressure. Information is also given on the proper amounts of carbon dioxide to compensate for the pressures developed within the sealed tubes by the various acid mixtures used.

HYDRATED CALCIUM SILICO-ALUMINATES

treatment if they have been ignited at high temperatures, dissolve rather readily in hydrochloric acid at temperatures ranging from 200° to 300° C. The treatment is applied in sealed glass tubes enclosed in steel pressure vessels charged with compressed carbon dioxide portland cement. In the Journal of Re-

search for December (RP1623), E. P. | HEATS OF COMBUSTION OF Flint and Lansing S. Wells describe the preparation of three new complex hydrated aluminates, as follows: (1) A low-silica calcium silicoaluminate, 3CaO. Al₂O₃CaSiO₃.12H₂O; (2) a high-silica calcium silicoaluminate of probable formula, 3CaO.Al2O3CaSiO30-32H2O; and (3) a hexacalcium aluminate, 3CaO. Al₂O₃.3Ca (OH)₂.30H₂O.

The first of these compounds is closely related to the low-sulfate calcium sulfoaluminate. 3CaO.Al2O2.CaSO4.12H2O. which may be formed during the setting of portland cement. The other two compounds are very similar to the highsulfate calcium sulfoaluminate, 3CaO. Al2O2.3CaSO4.31H2O, which is known to be a constituent of hydrated portland cement. Crystals of the high-silica silicoaluminate were indentified in mixtures of the following cement compounds which had stood in contact with approximately saturated lime solution for 3 years: β -2CaO.SiO₂ and 4CaO.Al₂O₃. Fe₂O₄, β -2CaO.SiO₂ and 3CaO.Al₂O₄, and 3CaO.SiO₂ and 4CaO.Al₂O₄Fe₂O₃. The low-silica silicoaluminate, on standing in contact with lime solution, is slowly transformed to the high-silica compound, and this behavior is analogous to that of the two forms of calcium sulfoaluminate.

The hydrated hexacalcium aluminate, 3CaO.Al2O3.3Ca (OH) 2.30H2O, probably does not occur as a direct hydration product of portland cement but might form, under certain conditions, as a solid solution in the high-sulfate form of

calcium sulfoaluminate.

HEAT OF FORMATION OF CARBON DIOXIDE

In the Journal of Research for December (RP1620), Edward J. Prosen, Ralph S. Jessup, and Frederick D. Rossini present a recalculation of data previously reported on the heats of comof graphite and diamond (TNB258, October 1938), together with some new data on graphite. As a result of this recent work, certain "best" values have been selected as representing the heats of combustion of graphite and diamond and the transition of graphite into diamond:

 $C(c, graphite) + O_2 (gas) = CO_2$ (gas); $\Delta H^{\circ}_{200,20} = -393,447 \pm 45$ int. 1/mole=94,051.8±10.8 cal/mole.

 $C(c, diamond) + O_2(gas) = CO_2(gas)$; $\Delta H^{\circ}_{100.15} = -395,343 \pm 96 \text{ int. j/mole} =$ -94,505.1 ±22.9 cal/mole.

C(c, graphite)=C(c, dlamond); ΔH° 286.16=1896 ±85 int. j/mole=453.2 ±20.3 cal/mole.

BENZOIC ACID AND GRAPHITE

Experimental data on the heats of combustion of benzoic acid and carbon in the form of graphite are presented by Edward J. Prosen and Frederick D. Rossini in the December Journal of Research (RP1619). The data on benzoic acid support the change in the value of its heat of combustion reported by Jessup two years ago (TNB306, October 1942). The data on graphite yield a slightly higher value for the heat of formation of carbon dioxide than that previously reported.

COMMERCIAL STANDARD FOR SOLID-FUEL-BURNING FORCED-AIR FURNACES

Commercial Standard CS109-44, now available in printed form is a recorded voluntary standard of the trade for solid-fuel-burning forced-air furnaces. It was developed by the industry in cooperation with the Government, on request of the Federal Housing Administration. The standard sets forth definitions, requirements for construction, performance under test, test code, informative labeling, and a uniform manufacturer's guarantee declaring the maximum and minimum output obtainable according to the commercial standard.

The standard will provide uniform methods of testing and rating, a Nationally recognized basis for guaranteeing performance and output, and will constitute a basis for fair competition. It became effective for new production on March 10, 1944.

Copies of CS109-44 are obtainable from the Superintendent of Documents, Government Printing Office, Washington 25, D. C., at 10 cents each.

PAINTING OF STRUCTURAL STEEL

Building Materials and Structures Report BMS102, recently released, gives the results of tests on a large number of priming and finish coat paints of various types to determine their durability and the degree of protection afforded to structural steel. This project, under the direction of Wilbur C. Porter, made use of accelerated laboratory and outdoor exposure tests. primings were applied to treated and untreated steel and galvanized iron panels. The effect of improving the protective value of primings by various pretreatment processes was given special consideration.

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The composition of the paints and the relative ratings, based on their performance in these tests, are discussed in detail. Many paints were found suitable for use in protecting metal structures. Zinc chromate and iron oxide-zinc chromate primings, when properly formulated with a synthetic resin vehicle, make very good rust-inhibitive coatings. They combine to a high degree the essential requirements necessary to withstand extremely severe corrosive atmospheres. The zinc dust paints, particularly the alkyd type, have good adhesion on galvanized iron. Among the topcoat finishes tested, aluminum, black, and dark-colored paints were the most durable.

This is the third and final report covering the investigation of surface treatment for corrosion protection of steel members for low-cost housing construction. Copies of BMS102 are obtainable from the Superintendent of Documents, Government Printing Office, Washington 25, D. C. The price is 10 cents.

COLORBLINDNESS TESTS

Through the committees of the Inter-Society Color Council, the Bureau has cooperated with the Army Air Forces, the U. S. Submarine Base at New London, and the Bureau of Medicine and Surgery, in trials and revisions of established tests of color vision and in the design of new tests. The purpose of this work is to assist the Navy and War Departments in their efforts to make sure that men with defective color vision are not placed where this defect would be a handicap.

About 2 percent of otherwise normal males lack the ability to distinguish redgreen differences, and to about half of this 2 percent, red signals not only fail to appear red but also have less than one-fifth normal brightness. Such observers are known as protanopes to distinguish them from other red-green-blind observers, to whom red signals are normally bright, known as deuteranopes. It has been found possible to express the quantitative data on the color confusions of protanople and deuteranopic observers, obtained by Pitt a few years ago in Great Britain, in terms of the ICI standard observer for colorimetry.

The derivation of the functions defining the color confusions of average red-green-blind observers is given in RP1618, Standard Response Functions for Protanopic and Deuteranopic Vision, by Deane B. Judd, in the December number of the Journal of Research. These functions have been useful in analyzing the performance of established tests for

color vision and in the design of new tests. This use of the functions is exemplified by the solution of three problems arising in design of colorbindness tests, and comparisons of these solutions are made with experimental results. It is pointed out that these functions accord with the form of three-components theory of color vision worked out in 1897 by Arthur König, and with the dominator-modulator form of that theory suggested in 1943 by Granit. These functions do not, however, form the basis for a complete theory of color vision, such as that of G. E. Müller.

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SHRINK-RESISTANT WOOL SOCKS

In order to increase the wear life of wool socks by decreasing shrinkage during laundry, the Quartermaster Corps requested that a project be initiated to study shrinkproofing methods for wool fabrics and knit goods. There was considerable difference of opinion as to the effect on wool of existing shrinkproofing methods, and the problem was to develop a treatment or to ascertain which of the existing methods would combine adequate reduction in shrinkage with minimum damage to the wool, and which, at the same time, would be readily adaptable to mill production. The fundamental research on various types of treatments was carried out by the Research Associateship of the Textile Foundation, and the adaptability of these treatments to mill production was tested by making regular mill runs at a pilot plant established in a large hosiery mill. As a result of the work, it was found that a treatment using calcium hypochlorite at a given pH cuts the shrinkages down to about 5 percent and meets all other requirements satis-This is in contrast to about 20- to 25-percent shrinkage of the untreated socks under the same conditions. Comparative wear tests, conducted under field conditions of wear and laundry, show that socks so treated can be expected to have a 60 percent longer life than the untreated socks. At present the treatment is being used on 60 to 70 percent of all "socks, wool, cushionsole," manufactured for the Army.

NEW AND REVISED PUBLICA-TIONS ISSUED DURING NOVEM-BER 1944

Journal of Research 2

Journal of Research of the National Bureau of Standards, volume 33, number 3, September 1944 (RP1602 to RP1605, inclusive). Price 30 cents. Annual subscription, 12 issues, \$3.50.

Research Papers 2

[Reprints from the September 1944 Journal of Research]

RP1602. Soil-corrosion studies, 1941: Ferrous and nonferrous corrosionresistant materials and nonbituminous coatings. Kirk H. Logan and Melvin Romanoff. Price 20 cents.

RP1603. Revised results obtained with certain dehydrating agents used for drying gases. John H. Bower. Price

5 cents.

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Bunber 1605,

mental equations of thermodynamics. Floyd Buckley. Price 10 cents.

Building Materials and Structures Reports 2

BMS102. Painting steel. Wilbur C. Porter. Price 10 cents.

Simplified Practice Recommendations 3

R6-44. Files and rasps (American pattern, and straight- and curved-tooth milled files). (Supersedes R6-42.) Price 10 cents.

Commercial Standards 2

CS109-44. Solid-fuel-burning forced-air furnaces. Price 10 cents.

² Send orders for publications under this heading only to the Superintendent of Documents. Government Printing Office, Washington 25. D. C. Subscription to Technical News Bulletin, 50 cents a year; Journal of Research, \$3.50 a year (to addresses in the United States and its possessions and to countries extending the franking privilege); other countries, 70 cents and \$4.50, respectively. tively.

Technical News Bulletin 3

Technical News Bulletin 331, November 1944. Price 5 cents. Annual subscription, 50 cents.

MIMEOGRAPHED MATERIAL

LETTER CIRCULARS

[Letter Circulars are prepared to answer specific inquiries addressed to the National Bureau of Standards and are sent only on request to persons having a definite need for the information. The Bureau cannot undertake to supply lists or complete sets of Letter Circulars or send copies automatically as issued. matically as issued.]

Transformation of the funda OC767. Inorganic analytical chemistry: Publications by members of the staff of the National Bureau of Standards.

(Supersedes LC675.)

LC770. Expanded table of weights per United States gallon and weights per cubic foot of sugar (sucrose) solutions at 20° C. Carl F. Snyder and Lester D. Hammond.

RECENT ARTICLES BY MEM-BERS OF THE BUREAU'S STAFF PUBLISHED IN OUTSIDE JOURNALS 3

The 6×42 binocular. Lieut. W. R. Bailey and G. M. Kline. Modern Plastics (122 East 42d Street, New York 17, N. Y.) 22, 105 (October 1944).

"Hotter than hell," not accurate measure of temperature. George D. Miller. The Taylor Meteor (Taylor Instrument Cos., Rochester 1, N. Y.) 8 No. 8, 2 (October 1944).

³These publications are not obtainable from the Government, unless otherwise stated. Requests should be sent direct to the publishers.

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